

# Developing ERICA FF2 for sulfate in cementitious systems

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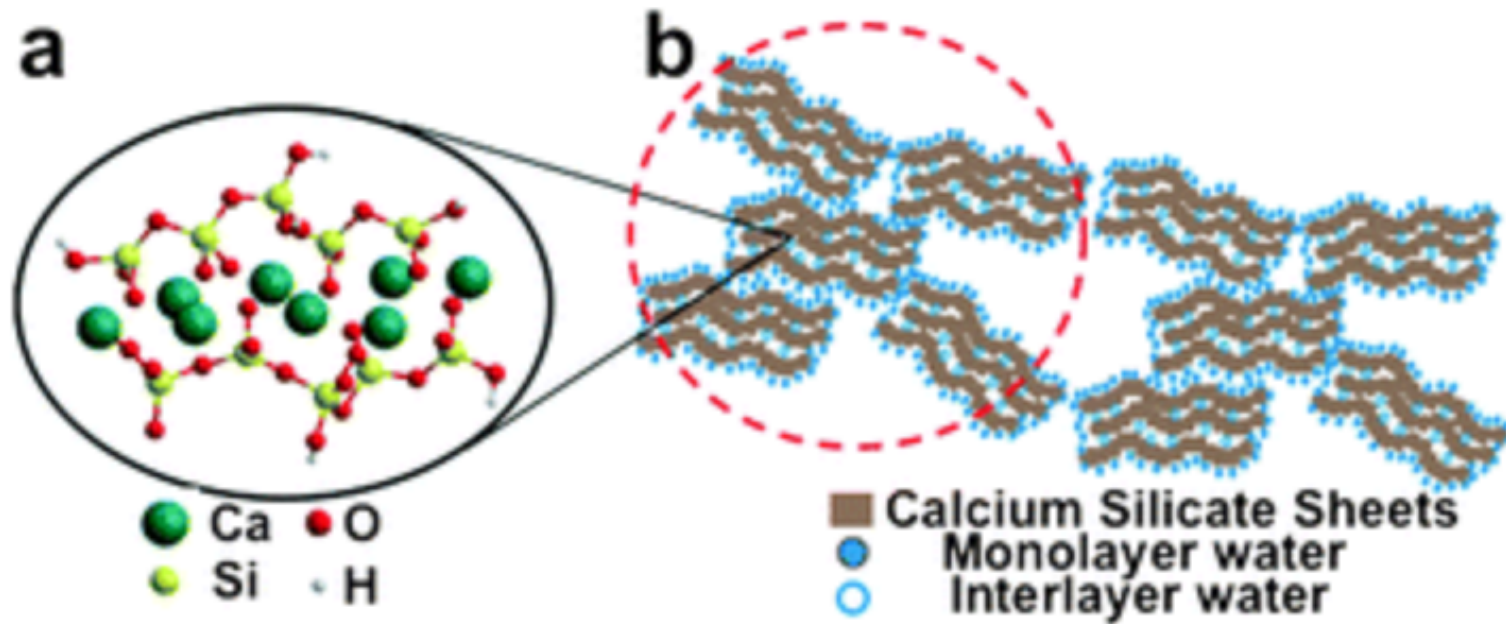
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- ESR3 goals
- Force field improvement and validation for Water-Sulfate-Portlandite solutions (ERICA FF1)
- Development of ERICA FF2 and Validation
- Future Plans
- Secondments

A lot of ions such as  $\text{SO}_4$

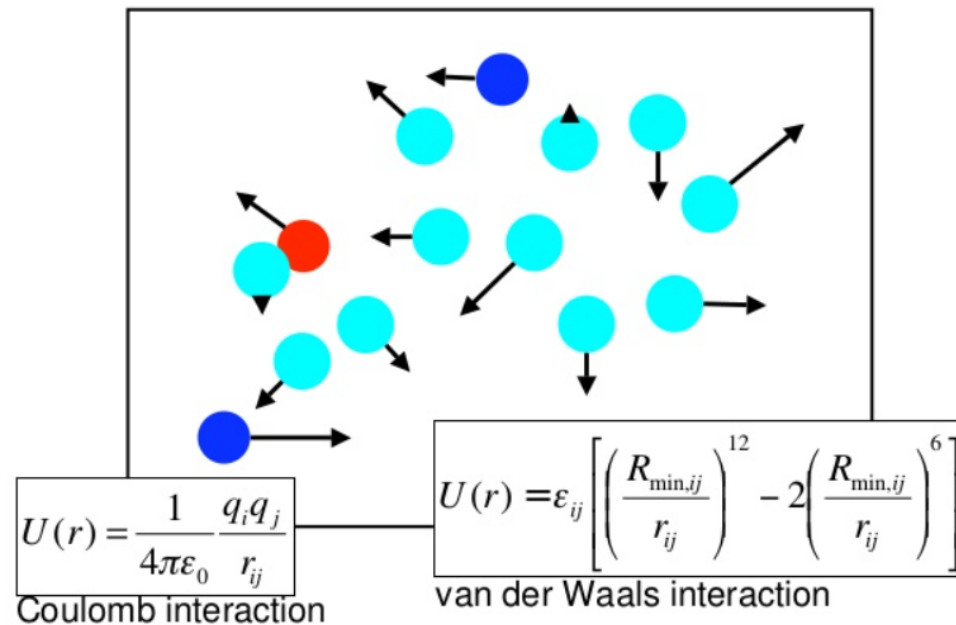


Goal: interaction of ions such as sulfate with surfaces of C-S-H

# ERICA FF1: Force Field Validation for Aqueous System of Portlandite and Sulfate

- ❑ Assembly of interatomic potentials
- ❑ Conclusions depending on the force field
- ❑ force field should be validated carefully

## Classical Molecular Dynamics



- ☐ The first step -the simulation of Portlandite growth in the presence of sulfate -develop an accurate empirical force field (FF).
- ☐ interaction of Portlandite- water, sulfate-water and sulfate-Portlandite must be captured correctly
- ☐ several force fields which can be used in the interactions of Portlandite-water :Clay FF and SPC/E,Cement FF and TIP4P2005
- ☐ Sulfate:Duvail(Polarizable) and Gale FF(non-polarizable)

☐ Gale model captured the structure of water around sulfate correctly

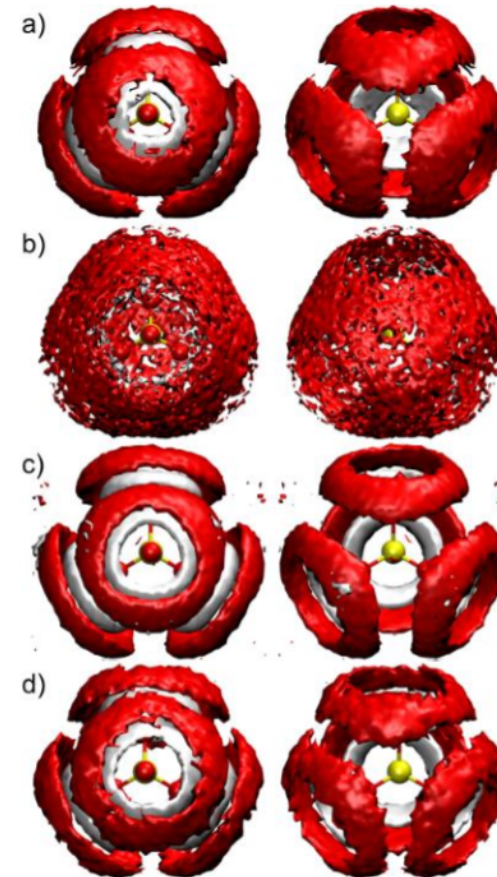
☐ A)FF1

☐ B)FF2

☐ C)AMOEBA

☐ D)DFT

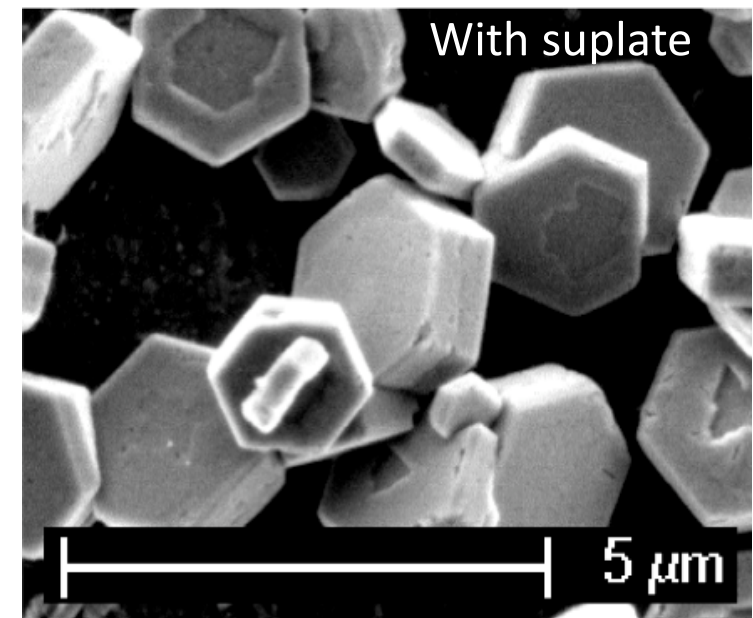
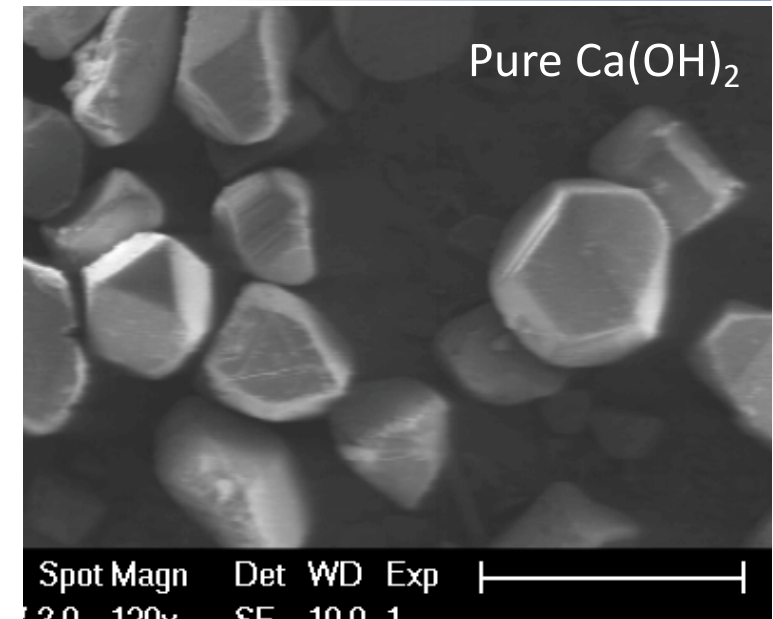
.Representations of the 3D water structure around the sulfate ion





- ❑ Force field developed during
- ❑ Sandra Galmarini thesis CEMENT FF1
- ❑ Does not have Al, Mg,  $\text{SO}_4^{2-}$
- ❑ Needs developments....
- ❑ Significant effect of sulfate on  $\text{Ca}(\text{OH})_2$  morphology\*
- ❑ Can influence porosity – hence transport properties & strength....
- ❑ Training for adsorption on C-S-H

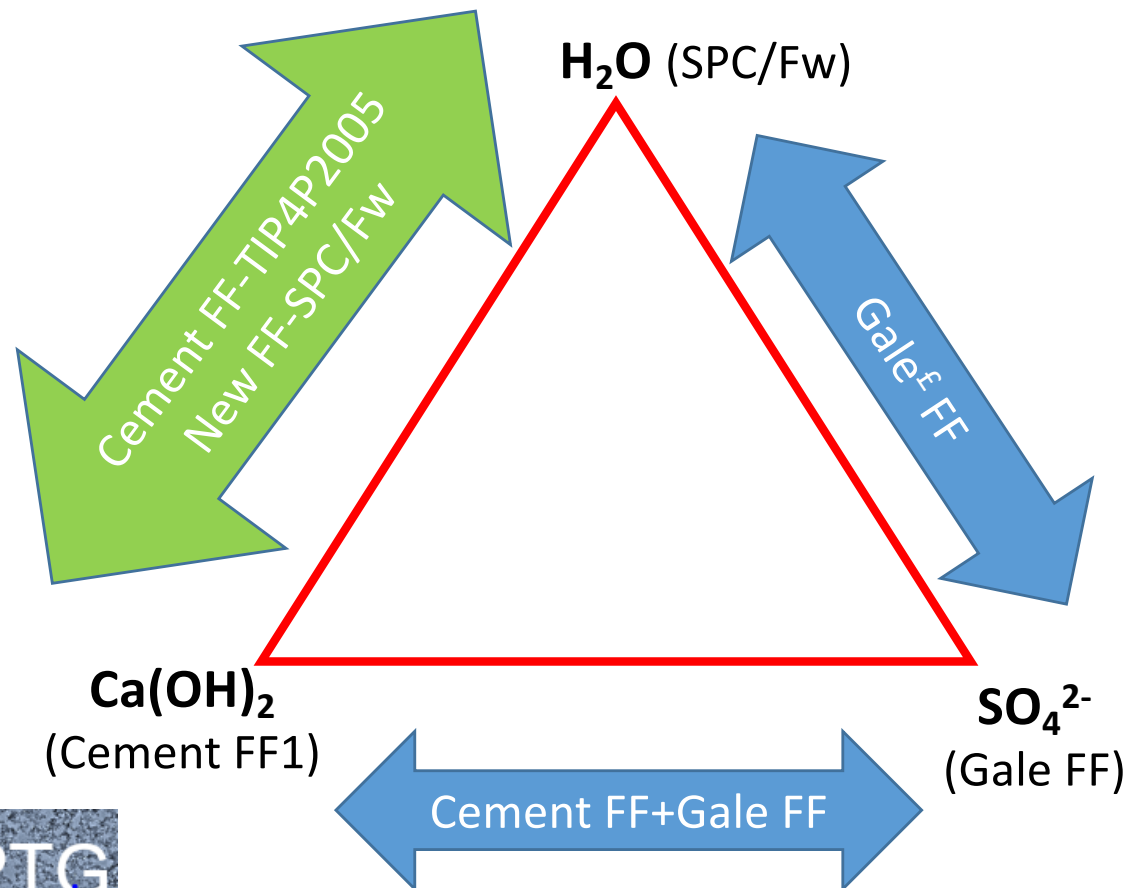
Galmarini et al CCR. 41 p1330 (2011) & CCR 71, p61 (2014).  
Galmarini & Bowen CCR 81,p16 (2016)





### Criteria for good force field

- %5-6 on the length parameters
- 10 degrees of angles



Development of non-polarizable force fields for phosphates and sulphates

£E. H. Byrne, P. Raiteri, and **J. D. Gale**,  
“Computational Insight into Calcium-Sulfate Ion Pair Formation,” *J. Phys. Chem. C*, 2017

- ☐ Experimental data for benchmarking performance of force field = a set of atomistic potentials

## Structural

- ☐ Portlandite –  $\text{Ca(OH)}_2$
- ☐ Gypsum –  $\text{Ca SO}_4 \cdot 2\text{H}_2\text{O}$
- ☐ Anhydrite –  $\text{Ca SO}_4$
- ☐ Coordination numbers
- ☐ Average bond lengths

## Thermodynamic

- ☐ Heats of dissolution

# ERICA FF2 development and validation

- Including Al and Si in the potential and interaction of these 2 atoms with other atoms
- Simulation of C-A-S-H is now accessible
- OH-SO4 validation for ERICA FF1 was missing-It has been done now by simulation of Ettringite

(J.am.Ceram soc,1999)**Nora de Leeuw FF**

**Cement FF 1**(cement and concrete research 2011,Sandra et al)

(J.phys.chem.c,2017)  
**Julian Gale FF**

**ERICA FF2**

**Cement FF 2**  
(cement and concrete research 2017,Aslam et al)

(J.Chemical Physics 2006)**SPC/Fw FF**



**Cement FF 3**(Aslam et al, Under preparation)



Ettringite Bulk simulation of a 2x2x1  
box

Total run of 1ns

1000 Atoms in the box

Figure in Z direction

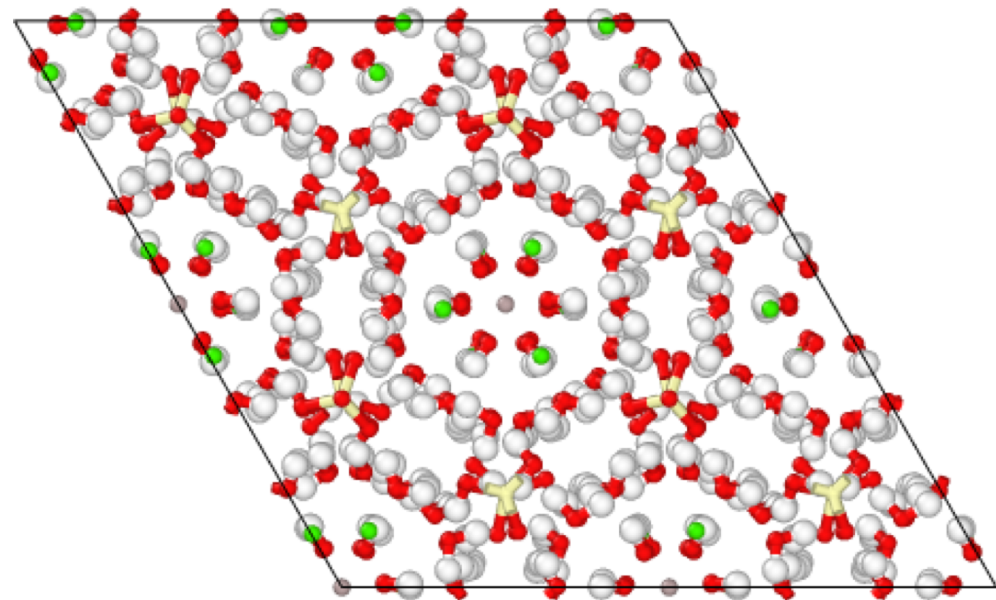
Grey: Aluminium

Green: Calcium

Yellow: Sulfur

Red: Oxygen

White: Hydrogen



	EM (A)	MD(A)	EXP(A)	Relative and absolute error EM	Relative and absolute error MD
a	11.47	11.67	11.23	2.1%	3.9%
b	11.38	11.63	11.23	1.3%	3.6%
c	21.67	21.6	21.48	0.8%	0.5%
alfa	90.34	89.88	90	0.34 abs	0.12 abs
beta	90.33	90.58	90	0.33 abs	0.58 abs
gamma	120.11	120.17	120	0.11 abs	0.17 abs

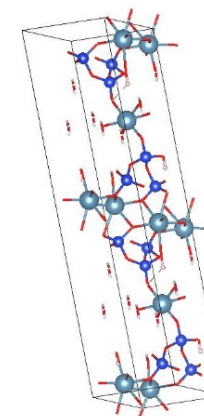
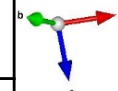


Average bond	MD	exp	relative error%
Ca-Ooh	2.52	2.53	0.35
Ca-Ow	2.31	2.41	4.14
Al-Ooh	1.85	1.79	3.35
S-Os	1.61	1.56	3.2
Ow-Hw	0.9976	1.03	3.1
Ooh-Hoh	0.9318	0.9756	4.0

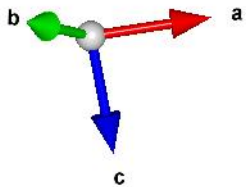
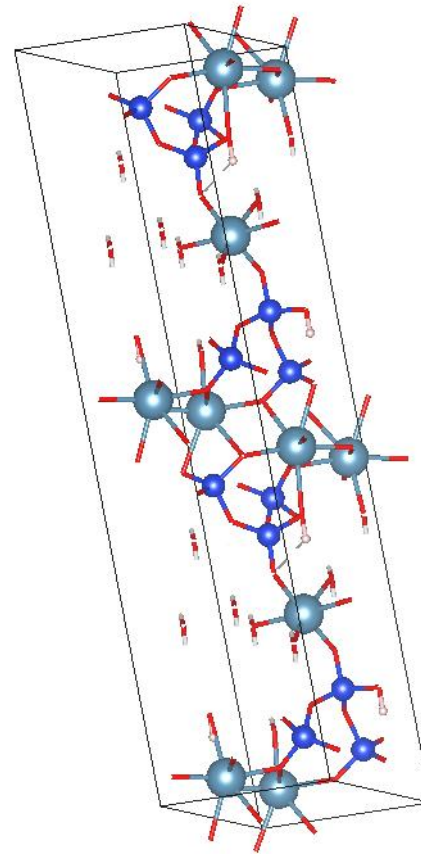
Tobermorite bulk simulation of a 4x4x1 box

Total run of 600 ps

	EM	MD	Exp	Relative error EM	relative error MD
a	6.75	6.81	6.73	0.001	0.002
b	7.37	7.39	7.42	0.004	0.003
c	31.11	31.62	30.65	0.011	0.015
alfa	110.48	112.47	111.34		
beta	87.04	87.12	87.37		
gamma	123.36	122.22	123.25		



# Why we are not happy by simulation of 14 Å Tobermorite





- Simulation of adsorption of sulfate into three surface of portlandite by MD and MTD
- Simulation of adsorption of sulfate into surfaces of C-S-H

- April 2020 2 weeks at Usurrey
- October 2020 1-2 months at CHRYSO

- Amirkabir University in Iran October 2019



